**2002 Vol. 4, No. 24 <sup>4209</sup>**-**<sup>4212</sup>**

**ORGANIC LETTERS**

## **Synthesis, Incorporation into Triplex-Forming Oligonucleotide, and Binding Properties of a Novel 2**′**-Deoxy-***C***-Nucleoside Featuring a 6-(Thiazolyl-5)benzimidazole Nucleobase**

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**Received July 25, 2002**

**ABSTRACT**



**6-(Thiazolyl-5)benzimidazole (Bt ) was designed as a novel nucleobase for the specific recognition of an inverted A**'**T base pair in a triple helix motif. It was successfully incorporated into an 18-mer triplex-forming oligonucleotide (TFO) using the 2**′**-deoxy-***C***-nucleoside phosphoramidite 16. The triple helix binding properties of the modified TFO were examined by means of thermal denaturation experiments targeting an oligopyrimidine**'**oligopurine 26-mer DNA duplex containing an A**'**T base pair inversion.**

Pyrimidine oligonucleotides can interact, in a parallel orientation, with the oligopurine strand of an oligopyrimidine' oligopurine DNA duplex through the formation of T'AxT and  $C$ <sup>+</sup> $GxC$ <sup>+</sup> $t$ riplets.<sup>1</sup> This recognition process occurs in the major groove of the double stranded DNA (ds-DNA) and involves specific Hoogsteen hydrogen bonds between pyrimidines of the triplex forming-oligonucleotide (TFO) and the Watson-Crick purine bases of its cognate ds-DNA duplex.2 Accordingly, TFOs have attracted considerable interest because of their potential applications in gene expression modulation and gene targeting technologies.<sup>3</sup> Unfortunately, when a single  $A \cdot T$  or  $G \cdot C$  base pair interrupts

an oligopyrimidine'oligopurine ds-DNA target, one observes a strong triplex destabilization.<sup>4</sup> Hence, to date, the recognition of mixed purine/pyrimidine sequences remains a challenge. In the past few years, many TFO chemical modifi-

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<sup>(1)</sup> The symbols ' and x indicate Watson-Crick and Hoogsteen hydrogen bonds, respectively.

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cations were reported to overcome this sequence limitation.5 The most promising approach consists of designing new modified bases able to form hydrogen bond contacts on both bases of the A<sup>-</sup>T or G<sup>-</sup>C Watson-Crick inverted base pairs in the major groove.<sup>6</sup> We have recently reported on the selective recognition of A'T base pair in ds-DNA by pyrimidine-motif TFO containing an unnatural nucleoside **S**. <sup>7</sup> This new base analogue consists of two unfused aromatic rings (3-aminophenyl-thiazole) linked to a 2-deoxyribose unit by an acetamide motif. This interesting result led us to design a nucleobase analogue featuring a thiazolyl-benzimidazole system (**Bt** ), to form selective hydrogen bonds with an inverted A'T base pair, as shown in Figure 1.



**Figure 1.** The proposed interactions between nucleobase **Bt** and the inverted **<sup>A</sup>**'**<sup>T</sup>** base pair.

**Bt** was designed on the basis of molecular modeling studies suggesting that a more rigid modified base would better stabilize triplex formation. Here we report on the synthesis of the new nucleobase **Bt** together with its incorporation into a TFO whose recognition properties have been thoroughly examined. To incorporate **Bt** into a TFO we needed to synthesize, as suggested by modelization, a 2′-deoxy-*C*nucleoside derivative of  $\mathbf{B}^t$  in which the deoxyribose unit is connected to the C2-position of the thiazolyl-benzimidazole aglycone **Bt** (Figure 1). A 2′-deoxy-C-nucleoside, in which the aglycone moiety is reduced to a single benzimidazole ring **B** (Figure 1), was also synthesized as a control.

A large number of synthetic approaches to a variety of C-nucleosides have been reported.8 In our case, the use of an aldol-type condensation between the 2-lithiated derivative of **5** and **6** with a suitably protected 2-deoxyribose appeared to be the most appropriate and straightforward procedure toward the preparation of **B**- and **Bt** -derived 2′-deoxy-*C*nucleosides. The key steps in this synthesis consisted of a Stille-type coupling, for the biaryl aglycone **Bt** formation (Scheme 1), and an aryl-aldol type condensation followed



<sup>*a*</sup> Reagents and conditions: (i) Boc<sub>2</sub>O, pyridine (100%); (ii)  $K_2CO_3$ , MeI, DMF (95%); (iii) LDA, THF,  $-78$  °C then Bu<sub>3</sub>SnCl (>95%); (iv) NaH, DMF, ClSO<sub>2</sub>NMe<sub>2</sub> (80%); (v) H<sub>2</sub>, Pd/C, MeOH/ THF  $(1/1)$  (>95%); (vi) TMSI, NaNO<sub>2</sub>, TEBAC, CH<sub>3</sub>CN (73%); (vii) Pd(PPh<sub>3</sub>)<sub>4</sub>, DMF, 60 °C (95%). \*Compound 6 was obtained from benzimidazole following conditions (iv).

by a regioselective isopropylidene cleavage-ring closure (Schemes 2 and 3).

In this study, the *N*,*N*-dimethylsulfamoyl and *tert*-butoxycarbonyl protecting groups were used for the respective protection of the benzimidazole and aminothiazole rings of **Bt** , and they were found to be compatible with oligonucleotide solid-phase synthesis.

First, the biaryl nucleobase **Bt** was obtained, according to our previously described procedure, by a palladiumcatalyzed Stille cross-coupling reaction between derivative **2** and 5(6)-iodobenzimidazole **4** in 95% yield (Scheme 1).10 Compound **2** was obtained from 2-aminothiazole **1** in three steps (>90% overall yield), which consisted of (i) standard (5) Reviews: (a) Sun, J. S.; He´le`ne, C. *Curr. Opin. Struct. Biol*. **<sup>1993</sup>**,

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treatment with  $Boc<sub>2</sub>O$ , (ii) methylation of the resulting carbamate, and (iii) stanylation at the thiazole C-5 position using LDA in THF, then trapping of the intermediate with tributyltin chloride. The 5(6)-iodo regioisomers **4**<sup>11</sup> derived from 5(6)-nitro-benzimidazole **3**: (i) benzimidazole protection (ClSO<sub>2</sub>NMe<sub>2</sub>/NaH/DMF), (ii) catalytic hydrogenolysis of the nitro function, and finally (iii) transformation of the obtained amine to the target iodo-derivatives  $4 \frac{(NaNO_2)}{2}$ TMSI)<sup>12</sup> in 73% yield (Scheme 1).

Next, and as shown in Scheme 2, the protected **Bt** and benzimidazole **B** were lithiated by LDA in THF at  $-50$  °C



*a* Reagents and conditions: (i) LDA, THF,  $-78$  °C then **7** (59– 61%); (ii) MsCl, Et3N, DMAP (88-90%).

to give the corresponding 2-lithio derivatives in quantitative yield.13 These 2-lithiated heterocycles were allowed to react with the known deoxyribose precursor **7**<sup>14</sup> to afford the corresponding pentitol derivatives **8** and **9** in 60% yield as a mixture of *R*/*S* diastereoisomers in 7/3 ratio, respectively, as determined by <sup>1</sup>H NMR. The diastereoisomers of each compounds **8** and **9** were then easily separated by silica gel flash chromatography.15 Mesylation of **8** and **9** with methanesulfonyl chloride afforded the 1-*O*-mesyl derivatives **10** (88%) and **11** (90%), respectively. To keep the required *N*,*N*dimethylsulfamoyl and *tert*-butoxycarbonyl protecting groups on the aglycone moiety for oligonucleotide solid-phase synthesis, we needed to use mild conditions to eliminate the isopropylidene group of compounds **10** and **11**. Interestingly, SnCl2 treatment of 1′-*O*-mesyl derivatives **10***S* and **11***S* in methylene chloride at room temperature resulted in a clean conversion to the cyclized product **12** and **13**, respectively, (60% yield) following a cleavage-ring closure process (Scheme 3).



*a* Reagents and conditions: (i)  $SnCl<sub>2</sub>·2H<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub> (58-62%)$ ; (ii) Bu<sub>4</sub>NF, THF (85-87%); (iii) DMTCl, Et<sub>3</sub>N, DMF (70-88%); (iv)  $CIP(OCH_2CH_2CN)(Pr)_2$ , Hunig's base,  $CH_2Cl_2$  (63–73%).

Under these mild conditions, the *S*-isomers of **10** and **11** led to the cyclized products  $\mathbf{12}(\beta)$  and  $\mathbf{13}(\beta)$ , respectively, whereas the  $R$ -isomers afforded the corresponding  $\alpha$ -anomers  $(12(\alpha)$  and  $13(\alpha)$ , respectively, see Supporting Information). In view of these data, it appears that pentitols **10** and **11** cyclize, under  $SnCl<sub>2</sub>$  treatment, according to an  $SN<sub>2</sub>$  process.<sup>16</sup> The  $\beta$ -anomeric configuration of 12 and 13 was assigned by 2D-NOESY experiments that showed, as expected for a *â*-configuration, a clear NOE correlation between H1′ and H4′. <sup>17</sup> The 5′-TBDPS group cleavage leading to the target **B**- and **Bt** -derived 2′-deoxy-*C*-nucleosides **14** and **15** was achieved by treatment of **12** and **13** with Bu4NF in THF. Finally, the preparation of the phosphoramidite building blocks **16** and **17**, which were used for TFO synthesis, was accomplished by successive 5′-dimethoxytritylation and 3′ phosphitylation of nucleosides **14** and **15**, respectively.

The phosphoramidites **16** and **17** were then incorporated into 18-mer TFOs (III) at the internal position Z (Figure 2) using automated oligonucleotide synthesis. The protected oligonucleotides thus obtained were purified by reverse phase HPLC, and the protecting DMT,  $SO_2NMe_2$  and Boc groups

<sup>(11)</sup> The iodo derivative **4** was obtained as a mixture of 5- and 6-regioisomers in nearly 40/60 ratio, respectively. Both regioisomers will give the same oligonucleotide after cleavage of the sulfamoyl protecting group. To facilitate the interpretation of the <sup>1</sup>H and <sup>13</sup>C NMR spectra, we only used the 6-isomer in the next steps of our synthesis.

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<sup>(15)</sup> The *R* and *S* configurations of compounds **8** and **9** were determined on the basis of the anomeric stereochemistry of their cyclized products **12** and **13**, respectively (NOESY experiments).

<sup>(16)</sup> Under more drastic conditions (TFA, H<sub>2</sub>O, 60 °C),<sup>14</sup> the cyclization occurred but with (i) concomitant cleavage of all the protecting TBDPS, SO2NMe2 and Boc groups and (ii) epimerization of the C-1′ stereocenter, which suggested a partial  $SN_2$ -type mechanism for the outcome of this reaction.

<sup>(17)</sup> In the same way, we observed for the other anomers  $(12(\alpha))$  and **13**( $\alpha$ ) obtained from **10***R* and **11***R*, respectively) clear NOE H1<sup>'</sup>-H3<sup>'</sup> and H1′-H5′ correlations, in accordance with an  $\alpha$  configuration.



**Figure 2.** Sequence of the studied triplexes I. IIxIII ( $X \cdot Y = A \cdot T$ and  $Z = G$ ,  $B^t$ , or  $B$ ) and melting temperature curves. Conditions:<br>I<sup>-II</sup> (1.2  $\mu$ M) III (1.5  $\mu$ M) in 10 mM cacodylate buffer (pH.6) I'II (1.2  $\mu$ M), III (1.5  $\mu$ M) in 10 mM cacodylate buffer (pH 6) containing 100 mM NaCl, 10 mM  $MgCl<sub>2</sub>$ , and 0.5 mM spermine.

were cleanly cleaved by treatment of the oligonucleotides with aqueous TFA solution (10%). To our knowledge, this is the first example of oligonucleotides synthesis in which *N*,*N*-dimethylsulfamoyl-protected nucleobases were involved.

The triple helix binding properties of these TFOs were examined by means of thermal denaturation experiments with the oligopyrimidine'oligopurine 26-mer ds-DNA (I'II) containing an A $\cdot$ T interruption (X $\cdot$ Y = A $\cdot$ T) (Figure 2).

We observed that the  $T<sub>m</sub>$  value obtained for the triplex containing an A<sup> $\cdot$ </sup>Tx**B**<sup> $\cdot$ </sup> triplet (*T*<sub>m</sub> = 43 °C) is very close to that obtained with **G**, which is known to form the most stable natural triplet (A $\cdot$ Tx**G**,  $T_m$  = 45 °C). However, as it is wellknown that **G** involves one hydrogen bond with **T** in A<sup> $\cdot$ TxG</sup> triplet (NH<sub>2</sub>(G)-O4 (T)),<sup>5c</sup> our **B<sup>t</sup>** nucleobase most likely does not involve three hydrogen bonds as illustrated in Figure 1. Interestingly, the replacement of  $B<sup>t</sup>$  by the benzimidazole nucleobase **B** caused a dramatic destabilization of the triplex ( $\Delta T_{\text{m}} = -9$  °C from A $\cdot$ Tx**B**<sup>t</sup> to A $\cdot$ Tx**B**) (Figure 2). This result shows clearly the implication of the methylaminothiazole moiety of **Bt** in the recognition process, which has not been yet elucidated.

To determine the specificity of the nucleobase **Bt** , the 18 mer TFO (III,  $Z = B<sup>t</sup>$ ) was also screened against duplexes<br>containing the three other possible base pairs  $(X \cdot Y = T \cdot A)$ containing the three other possible base pairs  $(X \cdot Y = T \cdot A$ , <sup>C</sup>'G and G'C) (Table 1). It was observed that (i) the





nucleobase **Bt** can moderately discriminate a Pu'Py from a Py'Pu base pair (A'Tx**Bt** and G'Cx**Bt** vs T'Ax**Bt** and  $C$ **·GxB<sup>t</sup>**) and (ii) that **B<sup>t</sup>** provides a +3 °C triplex stabilization<br>in the case of a G**·C** interruption compared to the best base in the case of a G'C interruption compared to the best base triplet made of the natural bases G'Cx**T**. The moderate discrimination of nucleobase **Bt** for all four base pairs argues for a non-sequence-specific interaction, i.e, third strand base stacking/intercalation.

In conclusion, we have synthesized a novel 2′-deoxy-Cnucleoside analogue featuring a highly functionnalized aglycone moiety (**B***<sup>t</sup>* ) and shown the *N*,*N*-dimethylsulfamoyl protecting group to be compatible with oligonucleotide synthesis. The new nucleobase was successfully incorporated into a TFO using automated oligonucleotide synthesis. The triplexes featuring an A'Tx**Bt** triplet or the most stable natural triplet A'Tx**<sup>G</sup>** were shown to be of comparable stability. In view of these results, **Bt** represents an interesting starting point for the synthesis of other nucleobases, for structureaffinity relationship studies which is currently in progress.

**Acknowledgment.** The authors thank M. Thomas for oligonucleotides synthesis, Dr. J.-C. Blais for characterization of oligonucleotides, and CNRS for a doctoral fellowship (BDI) to D.G.

**Supporting Information Available:** Experimental procedures and spectroscopic characterization (1H, 13C, 31P (for phosphoramidites) NMR and MS) of all new compounds. Also automated oligonucleotides synthesis, melting temperature experiments, and MALDI-TOF/MS characterization of oligonucleotides. This material is available free of charge via the Internet at http://pubs.acs.org.

OL026609H